

NEW DERIVATIVES OF BENZYLAMIDE WITH ANTICONVULSANT ACTIVITY

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Abstract: Previously obtained picolinic acid benzylamide is a potent anticonvulsant with low neurotoxicity. In search for new effective anticonvulsants twelve new benzylamides (**1-12**) were synthesized and preliminary evaluated in the Anticonvulsant Screening Program (ASP) of Antiepileptic Drug Development Program (ADDP) of NIH. Two of them appeared the most promising: 1-cyclopentenecarboxylic acid benzylamide (1-Cpc-BZA) (**9**) showed MES ED₅₀ = 85,36 mg/kg (PI = 2,49), scPTZ ED₅₀ = 1,37 mg/kg (PI = 1,37), 6Hz-EST ED₅₀ = 50,29 mg/kg and cyclopentanecarboxylic acid benzylamide (Cpc-BZA) (**11**) showed pilocarpine ED₅₀ = 154,75 mg/kg and pilocarpine ED₉₇ = 270,95 mg/kg.

Keywords: anticonvulsants, synthesis, activity, neurotoxicity

A number of aromatic amides of aromatic and heterocyclic acids have been synthesized in search for new antagonists of excitatory amino acids receptors with anticonvulsant activity (1 – 3). Some of them have been reported previously to be a potent agents. Generally, benzylamides were found to be more active than other amides. On the other hand, the most effective appeared amides of acids: picolinic, nicotinic, isonicotinic, nipecotic and isonipecotic. The most effective anticonvulsants came out to be picolinic acid benzylamide (Pic-BZA, PI against MES > 28.0) (1) and nicotinic acid benzylamide (Na-BZA, PI against MES = 4.70) (2). Some of derivatives of those compounds substituted in both rings were designed, prepared and pharmacologically evaluated. The best were: picolinic acid 2-fluorobenzylamide (Pic-2-F-BZA, PI against MES = 3.40) (3) and nicotinic acid benzylamide N-oxide (Nic-O-BZA, PI against MES < 5.6) (2).

This work was designed to extend the investigation to synthesis of derivatives of picolinic, nicotinic and isonicotinic acid benzylamides substituted with Cl, F and CF₃ in the rings. In the structure of compound **8** the distance between aromatic ring and amide group was prolonged with one CH₂ group to see the influence of this modification upon activity in comparison with picolinic acid benzylamide.

Also was attempted preparation of series of benzylamides of the acids containing 5- or 7-member rings in the structure. The following compounds were obtained: picolinic acid 2-chlorobenzylamide (Pic-2-Cl-BZA, **1**), picolinic acid 3-chlorobenzylamide (Pic-3-Cl-BZA, **2**), picolinic acid 4-chlorobenzylamide (Pic-4-Cl-BZA, **3**), picolinic acid 2-fluoro-3-trifluoromethylbenzylamide (Pic-2F-3-TFM-BZA, **4**), 2-chloronicotinic acid benzylamide (2-Cl-Na-BZA, **5**), 6-chloronicotinic acid benzylamide (6-Cl-Na-BZA, **6**), 2-chloroisonicotinic benzylamide (2-Cl-iNa-BZA, **7**), picolinic acid phenylethylamide, **8**, 1-cyclopentene-1-carboxylic acid benzylamide, 1-Cpc-BZA, **9**, 3-cyclopentene-1-carboxylic acid benzylamide (3-Cpc-BZA, **10**), cyclopentanecarboxylic acid benzylamide (Cpc-BZA, **11**), cyclohexanecarboxylic acid benzylamide (Chc-BZA, **12**). One of the prerequisites of the compounds activity is a correct lipophilicity of the molecule assuring a good biological barrier penetration. Therefore log P values of the partition coefficient between *n*-octanol and water of the designed compounds were taken under consideration. These values as a measure of lipophilicity were moderate for all synthesized compounds, that suggests a good penetration across biological barriers and therefore rather good activity.

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EXPERIMENTAL

Chemistry

General

All used acids (Pic, 2-Cl-Na, 6-Cl-Na, 2-Cl-iNa, 1-Cpc, 3-Cpc, Cpc and Chc) as well as isobutyl chloroformate were purchased from Aldrich. N-methylmorpholine and benzylamine were supplied from Merck. DMF and THF were from POCH Gliwice.

¹H NMR spectra were recorded on a Bruker DM 400 MHz spectrometer. Chemical shifts were measured as δ units (ppm) relative to tetramethylsilane. TLC was carried out on a 0.25 mm thickness silica gel plates (Merck Kieselgel 60 F-254). The spots were visualized in UV light or with 0.3% ninhydrin in EtOH (97:3). CC was carried out under gravity on silica gel (Merck, grade 230 – 400 mesh). The solvent system used in TLC and CC was CHCl₃/MeOH in different volume ratios. HPLC was performed on a chromatograph equipped with pump (Techma-Robot, Warsaw), UV detector LCD 2040 (Laboratory Přistroje, Praha) and a computer registrator/recorder (CHROMA POLLAB, Warsaw). The peaks were recorded at 210 nm. Elemental analyses were performed on a Perkin-Elmer Microanalyser. Melting points were determined in a Böetius apparatus.

Synthesis of amides

The compounds **1–12** were synthesized using the mixed anhydrides method (4) of peptide synthesis. Suitable acid (Pic, 2-Cl-Na, 6-Cl-Na, 2-Cl-iNa, 1-Cpc, 3-Cpc, Cpc or Chc) (10 mmol) was dissolved in DMF (15 mL) and THF (15 mL) was added. Next, N-methylmorpholine (10 mmol, 1.1 mL) was added and the mixture was stirred under nitrogen and chilled to -15°C. Isobutyl chloroformate (10 mmol, 1.3 mL) was added dropwise to keep the temperature below -15°C. Then, benzylamine (BZA) (10 mmol) in THF was added in small portions and the reaction mixture was stirred at -15°C for 30 min and at room temperature for 1 h. The solution was concentrated *in vacuo* and the residue was dissolved in EtOAc (20 mL). This solution was washed with 20 mL portions of 1 M HCl, saturated NaHCO₃ solution and saturated NaCl solution, then dried with anhydrous MgSO₄, filtered and concentrated *in vacuo*. The obtained compounds were purified by crystallization with MeOH/Et₂O, acetone/hexane and/or by CC (CHCl₃/MeOH, 95:5, v/v).

Computer calculations

The HyperChem 4.5 (Hypercube, Inc.) program was used. The semiempirical method PM 3 was applied for a single point calculation. Geometry

Table 1. Physical and analytical data of the synthesized compounds

No.	Compound	Formula	M.w.	Yield %	m.p. °C	R _f	log P
1	Pic-2-Cl-BZA	C ₁₃ H ₁₁ ON ₂ Cl	246.68	65	84-85	0.56 (A)	2.84
2	Pic-3-Cl-BZA	C ₁₃ H ₁₁ ON ₂ Cl	246.68	72	oil	0.61 (A)	2.84
3	Pic-4-Cl-BZA	C ₁₃ H ₁₁ ON ₂ Cl	246.68	65	83-85	0.65 (A)	2.84
4	Pic-2F-3-TFM	C ₁₄ H ₁₁ ON ₂ F ₄	298.23	58	96-97	0.76 (B)	3.35
5	2-Cl-Na-BZA	C ₁₃ H ₁₁ ON ₂ Cl	246.68	82	118-120	0.61 (C)	2.81
6	6-Cl-Na-BZA	C ₁₃ H ₁₁ ON ₂ Cl	246.68	83	112-113	0.64 (C)	2.81
7	2-Cl-iNa-BZA	C ₁₃ H ₁₁ ON ₂ Cl	246.68	78	99	0.49 (C)	2.81
8	Pic-PEA	C ₁₄ H ₁₄ ON ₂	226.27	88	oil	0.80 (A)	2.58
9	1-Cpc-BZA	C ₁₃ H ₁₅ ON	201.26	64	114-116	0.68 (B)	2.60
10	3-Cpc-BZA	C ₁₃ H ₁₅ ON	201.26	78	115-117	0.82 (B)	2.54
11	Cpc-BZA	C ₁₃ H ₁₇ ON	203.28	73	92-93	0.80 (B)	2.80
12	Chc-BZA	C ₁₅ H ₂₁ ON	231.35	71	117-118	0.76 (B)	3.60

The elemental analyses were within \pm 0.4% of the theoretical value. ¹H-NMR data clearly confirmed the proposed structure. Hydrophobicity of the compounds expressed as log P value was calculated by a computer method. HPLC purity of all compounds is 100%. A = CHCl₃/MeOH, 98:2, v/v; B = CHCl₃/MeOH, 99.5:0.5, v/v; C = CHCl₃/MeOH, 95:5, v/v. BZA = benzylamide; PEA = phenylethylamide; TFM = trifluoromethyl group; Pic = picolinic acid; Na = nicotinic acid; iNa = isonicotinic acid; Cpc = cyclopentanecarboxylic acid; 1-Cpc = 1-cyclopentene-1-carboxylic acid; 3-Cpc = 3-cyclopentene-1-carboxylic acid; Chc = cycloheptanecarboxylic acid.

Table 2. Preliminary pharmacological evaluation (ASP) of the synthesized compounds

Comp. no.	Mice <i>i.p.</i> Identification ASP no. 1 Test Class of ASP		Rats <i>p.o.</i> Identification ASP no. 2 Test Class of ASP			Anticonvulsant Identification ASP no. 8 Test (30 mg/kg) Class of ASP		
	MES	scPTZ	MES	scPTZ	TOX	MES	scPTZ	TOX
1	II	II	—	—	—	—	—	—
2	III	III	—	—	—	—	—	—
3	III	III	—	—	—	—	—	—
4	III	III	—	—	—	—	—	—
5	III	III	—	—	—	—	—	—
6	II	III	I	I	0	I	I	0
7	II	II	I	I	0	0	—	0
8	II	II	0	0	0	—	—	—
9	II	III	0	0	0	I	0	0
10	III	III	—	—	—	—	—	—
11	III	III	—	—	—	—	—	—
12	III	III	—	—	—	—	—	—

MES = maximal electroshock seizure test; PTZ = pentetetrazole; 0 = inactive or not neurotoxic at the dose used. Class I – anticonvulsant activity at a dose of 100 mg/kg or less, class II – anticonvulsant activity at a dose greater than 100 mg/kg and class III – no activity at a dose up to and including 300 mg/kg.

optimization was performed by a Polak-Ribiere algorithm. Afterwards, the QSAR Properties module using atomic parameters derived by Ghose et al. (5) was applied to calculate log P values as a measure of hydrophobicity of the optimized structures of the compounds.

Pharmacology

All synthesized compounds (**1-12**) were evaluated in mice in Identification ASP no. 1 Test according to the method described by Krall et al. (6). MES, scPTZ and TOX tests were performed after *i.p.* administration. On the basis of the findings, the compounds were included in one of three classes. Class I – anticonvulsant activity at a dose of 100 mg/kg or less, class II – anticonvulsant activity at a dose greater than 100 mg/kg and class III – no activity at a dose up to and including 300 mg/kg. (Table 2). Four selected compounds (**6, 7, 8, 9**) were evaluated in rats in Identification ASP no. 2 Test after *p.o.* administration at a dose of 30 mg/kg (Table 2) and three compounds (**6, 7, 9**) in rats in Anticonvulsant Identification ASP no. 8 Test after *i.p.* administration at a dose 30 mg/kg (Table 2). Both experiments were performed also according to the method of Krall et al. (6). The MES ED₅₀, scPTZ ED₅₀ and TOX TD₅₀ of the most promising compound (**9**)

were determined in mice in Quantification ASP no. 4 Test, according to the method reported by Swinyard et al. (7), after *i.p.* administration (Table 3). Activity of compound (**9**) was examined in mice in Anticonvulsant Evaluation ASP no. 7 Minimal Clonic Seizure (6 Hz) Test (6Hz-EST) by procedure described by Barton et al. (8) after *i.p.* administration (Table 3). Pharmacological evaluation of potential anticonvulsant activity of compounds **9** and **11** was accomplished using method of Racine (9) of pilocarpine induced seizures. Prevention of Pilocarpine Induced Status, ASP no. 71 Test (rats, *i.p.*, time 0 min.) and Pilocarpine-Induced Status, ASP no. 72 Test (rats, *i.p.*, time 30 min.) were performed (Table 3).

RESULTS

Physico-chemical data of the synthesized compounds are given in Table 1 and the results of preliminary pharmacological tests in Tables 2 and 3.

DISCUSSION

Of the compounds obtained in the present study Pic-2-Cl-BZA (**1**), 6-Cl-Na-BZA (**6**), 2-Cl-iNa-BZA (**7**) and Pic-PEA (**8**) demonstrated a potent

Table 3. Preliminary pharmacological evaluation (ASP) of the synthesized compounds

Compd.	Mice <i>i.p.</i> Quantification ASP no. 4	PI	Mice <i>i.p.</i> Anticonvulsant Evaluation Test ASP no. 7	Rats <i>i.p.</i> pilocarpine- induced status Test ASP no. 71	Rats <i>i.p.</i> Prevention of pilocarpine- induced status Test ASP no. 72 Test
9	MES ED ₅₀ (mg/kg) 85.36	ag. MES 2.49	6Hz-EST ED ₅₀ (mg/kg) 50.29	time 0 min. (mg/kg.) 300	time 30 min. (mg/kg) 200
	scPTZ ED ₅₀ (mg/kg) 154.74	ag. scPTZ 1.37	—	—	—
	TD ₅₀ (mg/kg) 212.60		—	—	—
11	Pilocarpine ED ₅₀ (mg/kg) 154.75		—	300	225
	Pilocarpine ED ₉₇ (mg/kg) 270.95		—	—	—
Valproic acid	MES ED ₅₀ (mg/kg) 425.1	ag. MES 0.6	—	—	—
	TD ₅₀ (mg/kg) 243.0		—	—	—
Pheno- barbital	MES ED ₅₀ (mg/kg) 9.1	ag. MES 6.7	—	—	—
	TD ₅₀ (mg/kg) 61.1		—	—	—

MES = maximal electroshock seizure test (60 Hz); 6Hz-EST = minimal electroshock seizure test; PI = TD₅₀ (against MES or scPTZ) / ED₅₀

anticonvulsant activity. However, their neurotoxicity was high (4/8 to 8/8 mice at a dose 100 mg/kg). Instead, the lowest neurotoxicity showed 1-Cpc-BZA (**9**) and Cpc-BZA (**11**) (both 0/8 mice at a dose 100 mg/kg). Among them compound **9** displayed stronger activity (3/3 mice at a dose 100 mg/kg), than compound **11** (1/3 mice at the same dose). MES ED₅₀ and scPTZ ED₅₀ of both are presented in Table 2. Because some clinically useful anticonvulsants can be poorly active in the standard MES and scPTZ tests but have anticonvulsant activity *in vivo*, activity of compound (**9**) was examined in Minimal Clonic Seizure (6 Hz) Test (6Hz-EST) (Table 3). Pharmacological evaluation of potential anticonvulsant activity of compounds **9** and **11** was accom-

plished in pilocarpine induced seizures. Prevention of Pilocarpine Induced Status showed both compounds at similar doses. Pilocarpine ED₅₀ and ED₉₇ of compound **11** were determined and are reported (Table 3). Pharmacological tests of the synthesized potential anticonvulsants seem to indicate that reduction of 6-member ring of the acidic fragment of the compound to 5-member one is leading to an increased activity and decreased neurotoxicity (**9**, **11**). On the contrary, enlargement of the ring to a 7-member one decreases anticonvulsant activity (**12**) (Table 3). Based on the reported results one can conclude that reduction of the compound molecular size caused by reduction of ring size might be advantageous assuring a better biological barriers

penetration. Pharmacological tests also showed, that enlargement of the distance between both rings in picolinic acid benzylamide (introduction of additional CH_2 group in picolinic acid phenylethylamide in compound **8**) reduces antiepileptic activity. This observation seems to show a decrease of receptor affinity of such structure.

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